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# A green HPLC technique with a 100% water mobile phase for detecting imidacloprid and its metabolite 6-chloronicotinic acid

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CHRONICLE	ABSTRACT					
Article history: Received July 21, 2015 Received in revised form August 29, 2015 Accepted 19 September 2015 Available online 20 September 2015	This paper describes a harmless HPLC technique for detecting imidacloprid (ICP) and its metabolite, 6-chloronicotinic acid (6CA), using an isocratic 100 % water mobile phase. Chromatographic separations were performed a Daisopak® SP-200-3-C1-P with water mobile phase and a photodiode-array detector. The total run time was < 6 min. The system suitability was well within the international acceptance criteria. The detection limits were 0.018 $\mu$ g/ml for ICP and 0.005 $\mu$ g/ml for 6CA, respectively. A harmless HPLC method for simultaneous detection CA methods and a photodiode area ware before a particulate the terms of terms of the terms of the terms of terms of terms of terms of the terms of					
Keywords: International harmonized analytical method Imidacloprid 6-Chloronicotinic acid 100 % water mobile phase HPLC	detecting ICP and 6CA was developed and may be further applied to the quantification in animal-derived foods.					
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#### 1. Introduction

Overuse of pesticides is always associated with risk of food residues. In December 2013, imidacloprid (1-(6-chloro-3-pyridylmethyl)-N-nitroimidazolidin-2-ylideneamine) (ICP) (Figure 1), a widely/frequently-used neonicotinoid insecticide, may affect the developing human nervous system, disclose the European Food Safety Authority. Experts from the authority propose that some guidance levels for acceptable exposure to ICP be lowered while further research is carried out to provide more reliable data on so-called developmental neurotoxicity.<sup>1</sup>

In pharmacokinetic findings of ICP in mammals, the two main metabolizing routes have been identified. The first is oxidative cleavage, yielding 6-chloronicotinic acid (6CA) (Fig. 1), which is conjugated with glycine to form a hippuric acid-type conjugate. These two metabolites together represented most of the identified metabolites. The metabolizing study in rats have found that the main <sup>\*</sup>Corresponding author.

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metabolites (about 30%) were 6CA and its glycine conjugate; all other biotransformation products were quantitatively of minor significance.<sup>2</sup>

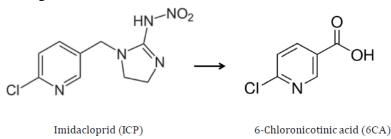


Fig. 1. Chemical structures of imidacloprid and its metabolite (6-Chloronicotinic acid)

To assure the safety of foods of animal origin for the consumer, the Codex sets maximum residue limits (MRLs) for the sum of ICP and its metabolites containing the 6-chloropyridinyl moiety, expressed as ICP.<sup>3</sup> Because determinations for ICP together with 6CA in the animal foods are therefore an important specific activity to guarantee food safety, a validated analytical method for the simultaneous determining ICP and 6CA is presently required.

In response to the recent expansion and diversification in the international food trade, the development of international harmonized methods to determine chemical residues in foods is essential to guarantee equitable international trade in these foods and ensure food safety for consumers. Whether in industrial nations or developing countries, an international harmonized method for residue monitoring in foods is urgently–needed. The ideal harmonized method must be easy-to-use, economical in time and cost, and must cause no harm to the environment and analyst.

Previous techniques based on high-performance liquid chromatographic (HPLC) combined with UV, photo-diode array (PDA), MS, and MS/MS for the monitoring ICP and 6CA<sup>4-6</sup> have crucial drawbacks: 1) all of the methods consume large quantities of toxic organic solvents, acetonitrile and/or methanol,<sup>7</sup> in the mobile phases. Risk associated with these solvents extend beyond direct implications for the health of humans and wildlife to affect our environment and the ecosystem in which we all reside. Eliminating the use of toxic solvents and reagents is an important goal in terms of environmental conservation, human health and the economy;<sup>8,9</sup> 2) most of the recent methods are based on LC-MS or -MS/MS. The facilities that LC-MS/MS system is available are limited to part of industrial nations because these are hugely expensive, and the methodologies use complex and specific. These are unavailable in a lot of laboratories for routine analysis, particularly in developing countries. No optimal method that satisfies the aforementioned requirements has yet been identified.

As the first examination problem in the establishment of an international harmonized method for the residue monitoring of ICP and 6CA, this paper describes an isocratic 100 % water mobile phase HPLC conditions to detect ICP and 6CA.

#### 2. Results and Discussion

#### 2.1 Optimum HPLC conditions

Using five types of C1 columns (Table 1), the author tested to achieve the separation with a 100 % water mobile phase. This study used water as an isocratic mobile phase and examined column temperatures  $\geq 25$  °C, the flow rates  $\geq 0.75$  ml/min, and HPLC retention times  $\leq 10$  min (Table 1). Because the HPLC separations were performed serially, the time/run was critical for routine residue monitoring. The short run time not only increased sample throughout for analysis but also affected the method-development time.

The five columns were compared with regard to 1) elution from the column; 2) separation between 6CA and ICP; 3) sharpness of peaks obtained upon injection of equal amounts. The chromatographic eluting and separating profiles within the conditions ranges examined are also presented in Table 1.

Under a 100% water mobile phase, the complete elution/separation of the two compounds and their sharp peaks were obtained by a Column-(C) with column temperature of 50 °C and flow rate of 1.0 ml/min. Fig. 2 displays that the resulting chromatogram obtained from the HPLC. The two target peaks are clearly distinguished at 1.40 and 5.30 min, respectively. The present HPLC-PDA analysis accomplished optimum separation in a short time without the need for a gradient system to improve the separation and pre-column washing after an analysis.

Table 1. Physical/chemical specifications of the C1 columns <sup>a</sup> used and chromatographic ICP and 6CA
separations obtained under the HPLC conditions examined <sup>b</sup>

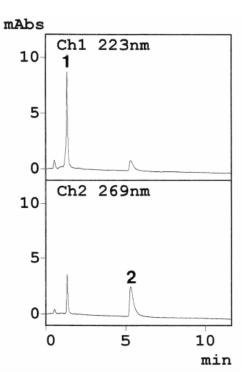
Column, Trade name	$d_{ m p}$	Pore diameter	Pore volume	Surface area	Carbon load	Chromatographic target peaks	
	(µm)	(nm)	(ml/g)	$(m^2/g)$	(%)	Elution & Separation	Peak form
(A) Developsil TMS-5	5	14	1.15	300	4.5	Not eluted <sup>c</sup>	
(B) Developsil TMS-3	3	14	1.15	300	4.5	Not eluted	
(C) Daisopak SP-200-3- C1-P	3	20	1.1	300	3	Eluted & Separated	Sharp
(D) Inertsil TMS	5	10	1.05	450	3.5	No ICP was eluted	
(E) Wakosil 5TMS	5	12	Nd <sup>d</sup>	Nd	4	Eluted & Separated	Broadening

<sup>a</sup> i.d.= 4.6 mm; length = 150 mm.

<sup>b</sup> Isocratic mobile phase of water; flow-rates  $\geq$  0.75 ml/min; column temperatures  $\geq$  25 °C; HPLC retention times  $\leq$  10 min.

<sup>c</sup> The column was difficult to elute 6CA and ICP peaks within 10 min (= the HPLC measuring time examined).

<sup>d</sup> No data.



**Fig. 2.** Typical chromatograms of a standard mixture (0.5 μg/ml) obtained from the HPLC system. PDA detector set at 223 nm (Ch 1) or 269 nm (Ch 2). The injection volume was 15 μl. Peaks, 1= 6CA (retention time, Rt = 1.40 min); 2= ICP (Rt = 5.30 min)

### 2.2 HPLC validation

Table 2 summarizes the validation data for the performance parameters. The linearity and system suitability values were well within the international acceptance criteria.<sup>10</sup>

	6CA <sup>a</sup>	<b>ICP</b> <sup>b</sup>	Acceptance criterion <sup>c</sup>
Linearity $(r)^d$	0.9997	0.9994	≥ 0.999
Range (µg/ml)	0.03	- 10	
Detection limit <sup>e</sup> (µg/ml)	0.005	0.018	
System suitability :			
a) Resolution (Rs)	5.7		>2
b) Injection repeatability <sup>f</sup> (RSD, %)			
Retentiontime	0.21	0.28	$\leq 1$
Peak area	0.25	0.42	$\leq 1$

**Table 2.** Chromatographic method validation data

<sup>b</sup> PDA detector set at 269 nm.

<sup>c</sup> Recommendations in the FDA guidelines.<sup>10</sup>

<sup>d</sup> r is the correlation coefficient (p < 0.01) for calibration curve.

<sup>e</sup> Detection limit as the concentration of analyte giving a signal-to-noise ratio = 3.

 $^{\rm f}$  Data as the relative standard deviations calculated for 10 replicate injections (10  $\mu l)$ 

of a mixed standard solution (0.5 µg/ml of ICP and 6CA, respectively).

#### 3. Conclusions

In the present study, a harmless HPLC-PDA method for detecting ICP and it metabolite 6CA using an isocratic 100 % water mobile phase has been successfully established. The water mobile phase method is harmlessness to the environment and to humans and has a short run time and high system suitability. The HPLC system may be proposed as an international harmonized method for detecting ICP and 6CA. For the quantification in animal-derived foods, the proposed HPLC method will be applicable enough by performing a suitable sample preparation technique.

## 4. Experimental

#### 4.1. Chemicals, Reagents, and Equipment

Standards of imidacloprid (ICP) and 6-chloronicotinic acid (6CA) (6-chloropyridine-3-carboxylic acid) and distilled water (HPLC grade) were purchased from Wako Pure Chem. Ltd. (Osaka, Japan).

The HPLC system, used for method development, included a model PU-980 pump and DG-980 50degasser (Jasco Corp., Tokyo, Japan) equipped with a model CO-810 column oven (Thosoh Corp., Tokyo, Japan), as well as a model SPD-M10A VP photodiode-array (PDA) detector (Shimadzu Scientific Instruments, Kyoto, Japan).

The following five types of C1 non-polar sorbent columns (3 or 5  $\mu$ m d<sub>P</sub>; 4.6 mm i.d.; 150 mm length) for HPLC analysis were used: Developsil® TMS-5 (or -3) (Nomura Chemical Co., Ltd., Aichi, Japan); Daisopak® SP-200-3-C1-P (Daiso Co., Ltd., Osaka, Japan); Inertsil® TMS (GL Sciences, Tokyo, Japan); Wakosil<sup>®</sup> 5TMS (Wako). Table 1 lists the particle physical specifications.

#### 4.2. Operating conditions

The analytical column was a Daisopak SP-200-3-C1-P ( $150 \times 4.6 \text{ mm}$ , 3 µm) column using an isocratic mobile phase of water at a flow rate of 1.0 ml/min at 50 °C. PDA detector was operated at 190 – 320 nm: the monitoring wavelengths were adjusted to 223 and 269 nm which represent maximums for 6CA and ICP, respectively. The injection volumes were  $10 - 20 \mu l$ .

## 4.3 Preparation of Stock Standards and Working Mixed Solutions

Stock standard solutions of ICP and 6CA were prepared by dissolving each compound in water followed by water to a concentration of 20  $\mu$ g/ml. Working mixed standard solutions of these two compounds were prepared by suitably diluting the stock solutions with water. These solutions were kept in a refrigerator (5 °C).

## 4.4 HPLC Validation

## 4.4.1 Linearity

The calibration curve was generated by plotting peak areas ranging from 0.03 to 10  $\mu$ g/ml versus their concentrations. The linearity was assessed from the linear regression with its correlation coefficient.

## 4.4.2 Detection limit

The detection limit should correspond to the concentration for which the signal-to-noise ratio. The value was defined as the lowest concentration level resulting in a peak area of three times the baseline noise.

## 4.4.3 System suitability test

The HPLC system suitability is an essential parameter of HPLC determination, and it ascertains the strictness of the system used. The suitability was evaluated as the relative standard deviations of peak areas and retention times calculated for 10 replicate injections of a mixed standard solution (0.5  $\mu$ g/ml).

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